Lesson Plan for CBCS system: Department of Chemistry 2021-22

Semeste	Progr	Course and	Topic	Teacher	No. Of
r	a -mme	Name of the Paper			hours
1	Hons	CC1-1-TH:	Extra nuclear Structure of atom	SB	14
		INORGANIC	Acid-Base reactions	SB+SG	6+6
		CHEMISTRY-1,	Redox reactions	SG	14
		ORGANIC	Basics of Organic Chemistry: Bonding	PR	18
		CHEMISTRY-	and Physical Properties		
		1A	General Treatment of Reaction	PR	2
			Mechanism I		
		CC1-1-P	INORGANIC CHEMISTRY: I (1) LAB: Acid	SB	30
			and Base Titrations and Oxidation-		
			Reduction Titrations		
			ORGANIC CHEMISTRY: O (1A) LAB:	PR	15
			Separation based upon solubility		
		Total n	umber of hours for CC 1-1 (Theory+Practical	al)	60T + 45P
	CC1-2-TH:	Kinetic Theory and Gaseous state	IS	20	
		PHYSICAL	Transport processes	IS	8
		CHEMISTRY-1, ORGANIC	Chemical kinetics	IS	12
			Stereochemistry I	PR	17
			General Treatment of Reaction Mechanism II	PR	3
			PHYSICAL CHEMISTRY: P (1) LAB	IS	30
		CC1-1-P	ORGANIC CHEMISTRY: O (1B) LAB:	PR	15
		CC1-1-F	Determination of boiling point of		
			common organic liquid compounds		
	Gen CC	Total n	umber of hours for CC 1-2 (Theory+Practical	al)	60T + 45P
		Gen CC1/GE1 TH	Kinetic Theory of Gases and Real gases	IS	7
			Liquids	IS	6
		Chemical Kinetics	IS	7	
			Atomic Structure	SB	7
			Chemical Periodicity	SG	7
			Acids and bases	SB+SG	3+3
			Fundamentalsof Organic Chemistry	PR	7
			Stereochemistry	PR	7
			Nucleophilic Substitution and Elimination Reactions	PR	6
		CC1/CF1 D		VCTCTC	45
		CC1/GE1 P	1. Estimation of sodium carbonate and	AS+IS+S G +MK	45
			sodium hydrogen carbonate present in a	J TIVIK	
			mixture.	1	

	T		1	T	_
		Total nu	2. Estimation of oxalic acid by titrating it with KMnO ₄ . 3. Estimation of water of crystallization in Mohr's salt by titrating with KMnO ₄ . 4. Estimation of Fe (II) ions by titrating it with K ₂ Cr ₂ O ₇ using internal indicator. 5. Estimation of Cu (II) ions iodometrically using Na ₂ S ₂ O ₃ . 6.Estimation of Fe(II) and Fe(III) in a given mixture using K ₂ Cr ₂ O ₇ solution.	ical)	60T +
2	Hons	CC-2-3-TH ORGANIC	Stereochemistry II	PR	45P 20
		CHEMISTRY-2	General Treatment of Reaction Mechanism III	PR	20
			Substitution and Elimination Reactions	PR	20
		CC-2-3-P	Organic Preparations, Purification and Melting point of the purified product	PR	45
		Total number of hours for CC 2-3 (Theory+Practical)			60T + 45P
		CC-2-4-TH	Chemical Bonding-I	SG	20
		INORGANIC	Chemical Bonding-II	SB	30
		CHEMISTRY-2	Radioactivity	SG	10
		CC-2-4-P	lodo-/ lodimetric Titrations, Estimation of metal content in some selective samples	SB	45
		Total number o	f hours for CC 2-4 (Theory+Practical)	•	60T + 45P
	Gen	CC/GE 2 TH	Chemical Thermodynamics	IS	8
			Chemical Equilibrium:	IS	7
			Solutions	Is	5
			Phase Equilibria	IS	5
			Solids	IS	5
			Aliphatic Hydrocarbons	PR	10
			Error Analysis and Computer Applications	SG	10
			Redox reactions	SG	10
		CC/GE 2 P	 Study of kinetics of acid-catalyzed hydrolysis of methyl acetate Study of kinetics of decomposition of H₂O₂ (Clock Reaction) Study of viscosity of unknown liquid (glycerol, sugar) with respect to water. 	AS+IS+S G +MK	45
	<u> </u>		(Brycerol, Sugar) with respect to water.		

3	HONS	CC-3-5-TH PHYSICAL	4. Determination of solubility of sparingly soluble salt in water, in electrolyte with common ions and in neutral electrolyte (using common indicator) 5. Preparation of buffer solutions and find the pH of an unknown buffer solution by colour matching method 6. Determination of surface tension of a liquid using Stalagmometer fhours for CC-2/GE-2 (Theory+Practical) Chemical Thermodynamics I Chemical Thermodynamics II	IS IS	60T + 45P 10 20
		CHEMISTRY-2	Applications of Thermodynamics – I	Is	6
			Electrochemistry	AS	24
		CC-3-5-P	1. Conductometric titration of an acid (strong, weak/ monobasic, dibasic, and acid mixture) against strong base. 2. Study of saponification reaction conductometrically 3. Verification of Ostwald's dilution law and determination of Ka of weak acid 4. Potentiometric titration of Mohr's salt solution against standard K ₂ Cr ₂ O ₇ and KMnO ₄ solution 5. Determination of Ksp for AgCl by potentiometric titration of AgNO ₃ solution against standard KCl solution 6. Determination of heat of neutralization of a strong acid by a strong base		45
		Total number o	f hours for CC 3-5 (Theory+Practical)		60T +
				ı	45P
		CC-3-6-TH	Chemical periodicity	SG	15
		INORGANIC CHEMISTRY-3	Chemistry of s and p Block Elements	SG	15
			Noble Gases, Inorganic Polymers	SB	15
			Coordination Chemistry-I	SB	15
		CC-3-6-P	Complexometric titration, Chromatography of metal ions, Gravimetry	SG	45
		Total number o	f hours for CC 3-6 (Theory+Practical)	•	60T + 45P
		CC-3-7-TH	Chemistry of alkenes and alkynes	PR	15

	ORGANIC	Carbonyl and Related Compounds	PR	30
	CHEMISTRY-3	Organometallics	PR	5
	CC-3-7-P	A. Identification of a Pure Organic Compound Solid compounds: oxalic acid, tartaric acid, citric acid, succinic acid, resorcinol, urea, glucose, cane sugar, benzoic acid and salicylic acid Liquid Compounds: formic acid, acetic acid, methyl alcohol, ethyl alcohol, acetone, aniline, dimethylaniline, benzaldehyde, chloroform and nitrobenzene	PR	45
		B. Quantitative Estimations of glycine, glucose, sucrose, aniline, acetic acid in vinegar, urea, saponification value of oil.		
	Total number of	f hours for CC 3-7 (Theory+Practical)		60T + 45P
	SEC(A)-3-2-TH	Carbohydrates, Proteins, Enzymes	AS	10
	ANALYTICAL CLINICAL	Lipids, Lipoproteins	AS	10
	BIOCHEMISTR Y	Biochemistry of disease: A diagnostic approach by blood/ urine analysis	As	10
	Total number of	hours for SEC(A)-3-2-TH		30T
Gen	CC/GE 3 TH	Chemical Bonding : Ionic and covalent bonding	SG	10
		Chemical Bonding: MO Approach	SB	5
		Comparative study of p-block elements	SG	5
		Transition Elements (3d series and Lanthanoids and actinoid)	SB	5
		Coordination Chemistry	SB	5
		Electrochemistry: Ionic Equilibria		5
		Conductance, Electromotive force		10
		Aromatic Hydrocarbons, Organometallic Compounds, Aryl Halides	PR	15
	CC 3/GE 3 P	Qualitative semimicro analysis of mixtures containing two inorganic radicals.	SB+AS+IS +MK	45
	Total number of	f hours for CC-3/GE-3 (Theory+Practical)	•	60T + 45P
	SEC(A)-3-1-TH Basic Analytical	Introduction to Analytical Chemistry, Chromatography, Ion-exchange, Suggested Applications, Suggested Instrumental demonstrations	SG	15
	Chemistry	mstrumental demonstrations		

		Total number o	Analysis of soil, Analysis of water, Analysis of food products, Analysis of cosmetics f hours for SEC(A)-3-1-TH	IS	15 30T
4	HONS	CC-4-8-TH	Nitrogen compounds	PR	12
-	110113	ORGANIC	Rearrangements	PR	14
		CHEMISTRY-4	The Logic of Organic Synthesis	PR	14
		CHEWISTKI	Organic Spectroscopy	PR	20
		CC-4-8-P	Qualitative Analysis of Single Solid	PR	45
		CC-4-6-F	Organic Compounds	FIX	45
		Total number o	f hours for CC 4-8 (Theory+Practical)		60T + 45P
		CC-4-9-TH PHYSICAL	Application of Thermodynamics – II: Colligative properties	IS	10
		CHEMISTRY-3	Phase Equilibrium	IS	10
			Foundation of Quantum Mechanics	AS	25
			Crystal Structure	AS	15
		CC-4-9-P	 Kinetic study of inversion of cane sugar using a Polarimeter (Preferably Digital) Study of Phase diagram of Phenol-Water system. Determination of partition coefficient for the distribution of I2 between water and CCl₄ Determination of pH of unknown solution (buffer), by colour matching method pH-metric titration of acid (monoand di-basic) against strong base pH-metric titration of a tribasic acidagainst strong base. 	IS	45
		Total number o	f hours for CC 4-9 (Theory+Practical)		60T + 45P
		CC-4-10-TH	Coordination Chemistry-II	SB	22
		INORGANIC CHEMISTRY-4	d-d transitions; L-S coupling; qualitative Orgel diagramscharge transfer spectra	SG	8
			Chemistry of d- block elements	SB	8
			Chemistry of f- block elements	SG	7
			Inorganic Reaction Kinetics and Mechanism	SG	15
		CC-4-10-P	Inorganic preparations, Instrumental Techniques: 1. Measurement of 10Dq by spectrophotometric method. 2. Determination of λ _{max}	SG	45

		Total number of	f hours for CC 4-10 (Theory+Practical)		60T + 45P
		SEC(B)-4-1-TH PHARMA- CEUTICALS	Drugs & Pharmaceuticals, Drug discovery, design and development; Basic Retrosynthetic approach.	PR	10
		CHEMISTRY	Synthesis of the representative drugs of the following classes: analgesics agents, antipyretic agents, antiinflammatory agents; antibiotics; antibacterial and antifungal agents; antiviral agents, Central Nervous System agents, Cardiovascular, antilaprosy, HIV-AIDS related drugs.	PR	10
			Fermentation		10
		Total number of	f hours for SEC(B)-4-1-TH		30T
	GEN	CC 4/GE 4 TH	Alcohols, Phenols and Ethers, Carbonyl Compounds, Carboxylic Acids and Their Derivatives, Amines and Diazonium Salts, Amino Acids and Carbohydrates	PR	20
			Crystal Field Theory	SB	20
			Quantum Chemistry & Spectroscopy		20
		CC 4/GE 4 P	1.Qualitative Analysis of Single Solid Organic Compound 2. Identification of a pure organic compound	SB+AS+IS +MK	45
		Total number of	f hours for CC-4/GE-4 (Theory+Practical)		60T + 45P
		SEC(B)-4-3-TH PHARMA- CEUTICALS CHEMISTRY	Drugs & Pharmaceuticals Drug discovery, design and development; Basic Retrosynthetic approach.	PR	10
			Synthesis of the representative drugs	PR	10
			Fermentation	SG	10
		Total number of	f hours for SEC(B)-4-1-TH	1	30T
5	HONS	CC-5-11-TH	Quantum Chemistry II	AS	30
		PHYSICAL	Statistical Thermodynamics	AS	20
		CHEMISTRY-4	Numerical Analysis	IS	10
		CC-5-11-P	Computer programs(Using FORTRAN or C or C ++) based on numerical methods	AS	45
		Total number of	f hours for CC-5-11 (Theory+Practical)		60T + 45P
		CC-5-12-TH	Carbocyles and Heterocycles	PR	16

	ORGANIC	Cyclic Stereochemistry	PR	10
	CHEMISTRY-5	Pericyclic reactions	PR	8
		Carbohydrates	PR	14
		Biomolecules	PR	12
	CEMA-CC-5-	A. Chromatographic Separations	Pr	45
	12-P	B. Spectroscopic Analysis of Organic		
		Compounds		
	Total number of	f hours for CC-5-12 (Theory+Practical)		60T +
				45P
	DSE(A)-5-2-TH	Computer Programming Basics	AS	20
	APPLICATIONS	(FORTRAN)		
	OF	Introduction to Spreadsheet	AS	25
	COMPUTERS	Software(MS Excel)		
	IN CHEMISTRY	Statistical Analysis	IS	15
	DSE(A)-5-2-P	Applications of computers in chemistry	AS	45
	Total number of	f hours for DSE(A)-5-2 (Theory+Practical)		60T +
				45P
	DSE(B)-5-1-TH	Silicate Industries	SB	16
	INORGANIC	Fertilizers	SB	8
	MATERIALS OF	Surface Coatings	SG	10
	INDUSTRIAL	Batteries	SB	6
	IMPORTANCE	Alloys	SG	10
		Catalysis	SG	6
		Chemical explosives	SG	4
	DSE(B)-5-1-P	1 . Determination of free acidity in	SB	45
		ammonium sulphate fertilizer.		
		2. Estimation of phosphoric acid in		
		superphosphate fertilizer.		
		3. Determination of composition of		
		dolomite (by complexometric titration).		
		4. Analysis of (Cu, Ni); (Cu, Zn) in alloy		
		or synthetic samples.		
	Total number of	5. Analysis of Cement.		COT
	lotal number of	f hours for DSE(B)-5-1 (Theory+Practical)		60T + 45P
GEN	DSE(A)-5-2-TH	Silicate Industries	SB	16
	INORGANIC	Fertilizers	SB	8
	MATERIALS OF	Surface Coatings	SG	10
	INDUSTRIAL	Batteries	SB	6
	IMPORTANCE	Alloys	SG	10
		Catalysis	SG	6
		Chemical explosives	SG	4
	DSE(A)-5-2-P	1 . Determination of free acidity in	SG+MK	45
		ammonium sulphate fertilizer.		
		2. Estimation of phosphoric acid in		
		superphosphate fertilizer.		

			3. Determination of composition of		
			dolomite (by complexometric titration).		
			4. Analysis of (Cu, Ni); (Cu, Zn) in alloy		
			or synthetic samples.		
		Table 1	5. Analysis of Cement.	1	COT
		Total number o	f hours for DSE(A)-5-2 (Theory+Practical)		60T + 45P
		SEC(A)-5-2-TH	Carbohydrates, Proteins, Enzymes	AS	10
		ANALYTICAL	Lipids, Lipoproteins	PR	10
		CLINICAL	Biochemistry of disease: A diagnostic	SG	10
		BIOCHEMISTR Y	approach by blood/ urine analysis		
			f hours for SEC(A)-5-2-TH	ı	30T
6	HONS	CC-6-13-TH	Theoretical Principles in Qualitative	SB	5
		INORGANIC	Analysis of cations		
		CHEMISTRY-5	Theoretical Principles in Qualitative	SG	5
			Analysis of anions		
			Bioinorganic Chemistry	SB	25
			Organometallic Chemistry	SG	25
		CC-6-13-P	Qualitative semimicro analysis of	SB	45
		CC-0-13-P	•	36	45
			mixtures containing not more than		
			three		
		-	radicals.		COT
		Total number o	f hours for CC-6-13 (Theory+Practical)		60T +
		00.011.70	Tana a s	T	45P
		CC-6-14-TH	Molecular Spectroscopy	IS	25
		PHYSICAL	Photochemistry and Theory of reaction	IS	15
		CHEMISTRY-5	rate		
			Surface phenomenon	IS	15
			Dipole moment and polarizability	Is	5
		CC-6-14-P	1. Determination of surface tension of a	AS	45
			liquid using Stalagmometer		
			2. Determination of the indicator		
	1	1			i i
			constant of an acid base indicator		
			constant of an acid base indicator spectrophotometrically		
			spectrophotometrically		
			spectrophotometrically 3. Verification of Beer and Lambert's		
			spectrophotometrically 3. Verification of Beer and Lambert's Law for $KMnO_4$ and $K_2Cr_2O_7$ solution		
			spectrophotometrically 3. Verification of Beer and Lambert's Law for KMnO ₄ and K ₂ Cr ₂ O ₇ solution 4. Study of kinetics of K ₂ S ₂ O ₈ + KI		
			spectrophotometrically 3. Verification of Beer and Lambert's Law for KMnO ₄ and K ₂ Cr ₂ O ₇ solution 4. Study of kinetics of K ₂ S ₂ O ₈ + KI reaction, spectrophotometrically		
			spectrophotometrically 3. Verification of Beer and Lambert's Law for KMnO ₄ and K ₂ Cr ₂ O ₇ solution 4. Study of kinetics of K ₂ S ₂ O ₈ + KI reaction, spectrophotometrically 5. Determination of pH of unknown		
			spectrophotometrically 3. Verification of Beer and Lambert's Law for KMnO ₄ and K ₂ Cr ₂ O ₇ solution 4. Study of kinetics of K ₂ S ₂ O ₈ + KI reaction, spectrophotometrically 5. Determination of pH of unknown buffer, spectrophotometrically		
			spectrophotometrically 3. Verification of Beer and Lambert's Law for KMnO ₄ and K ₂ Cr ₂ O ₇ solution 4. Study of kinetics of K ₂ S ₂ O ₈ + KI reaction, spectrophotometrically 5. Determination of pH of unknown buffer, spectrophotometrically 6. Determination of CMC of a micelle		
		Total number of	spectrophotometrically 3. Verification of Beer and Lambert's Law for KMnO ₄ and K ₂ Cr ₂ O ₇ solution 4. Study of kinetics of K ₂ S ₂ O ₈ + KI reaction, spectrophotometrically 5. Determination of pH of unknown buffer, spectrophotometrically 6. Determination of CMC of a micelle from Surface Tension Measurement.		607
		Total number o	spectrophotometrically 3. Verification of Beer and Lambert's Law for KMnO ₄ and K ₂ Cr ₂ O ₇ solution 4. Study of kinetics of K ₂ S ₂ O ₈ + KI reaction, spectrophotometrically 5. Determination of pH of unknown buffer, spectrophotometrically 6. Determination of CMC of a micelle		60T + 45P

		NATURAL PRODUCT	Alkaloids	PR	5
		OF	Future Trends in Green Chemistry	PR	12
		AND CHEMISTRY	Examples of Green Synthesis/ Reactions and some real world cases	PR	20
		GREEN CHEMISTRY	Principles of Green Chemistry and Designing a Chemical synthesis	PR	16
G	GEN	DSE(B)-6-1-TH	Introduction to Green Chemistry	PR	4
		Total number of	f hours for DSE(B)-6-4 (DISSERTATION)		105
			biological applications		
			complexes: Synthesis, types and		
			6. An overview on Schiff bases and its	1	
			diagram of metal carbonyl catalysts	30	103
			4. 5. A brief overview on energy profile	SG	105
			3.	IS	105
					105
			2.	1	
		DISSERTATION		1.5	
		DSE(B)-6-4-TH	1.	AS	45P 105
		Total number of	f hours for DSE(A)-6-3 (Theory+Practical)		60T +
		DSE(A)-6-3-P	Some green synthesis	PR	45
		PRODUCT 3	Terpenes	PR	3
		NATURAL PRODUCT S	Alkaloids	PR	5
		OF	Future Trends in Green Chemistry	PR	12
		CHEMISTRY	and some real world cases		
		AND	Examples of Green Synthesis/ Reactions	PR	20
		GREEN CHEMISTRY	Principles of Green Chemistry and Designing a Chemical synthesis	PR	16

PR: Dr. Priyabrata Roy

AS: Dr. Anuva Samanta

IS: Dr. Ishita Saha

MK: Smt. Malati Kundu

SG: Dr. Soumavo Ghosh